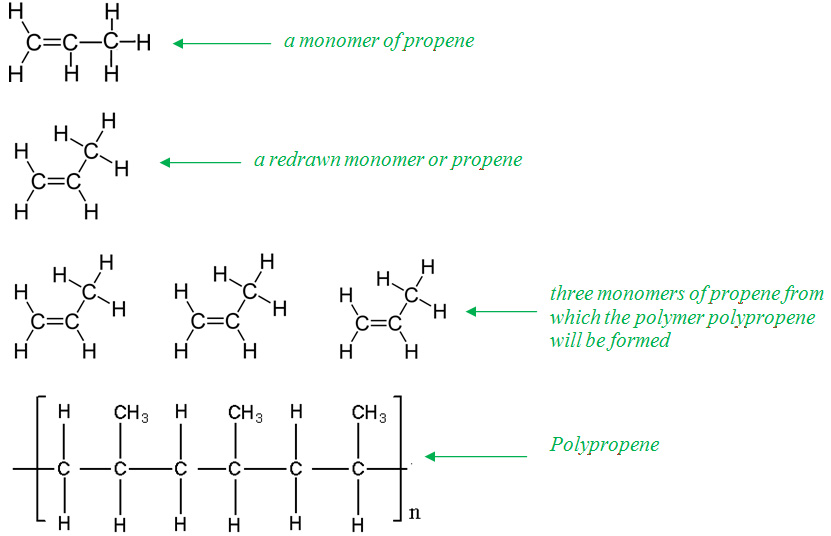
**Polymerisation (Level 2) examiners tips: Read these please!**

• when forming polymers (poly means many) break the double bond between the carbon atoms

• remember the brackets and “n” indicating that the polymer can be any length

• be careful when drawing polypropene, a common error is to form a polymer which is in fact polyethene!  
• do practice questions to recognise a monomer (mono means one) used to make the polymer



**Also…”don’t be daft”**don’t forget that the monomer DOES have a DOUBLE bond between the carbon atoms

don’t draw a carbon to carbon double bond in the POLYMER

you’ll need double BRACKETS and a little n shown for the polymer

remember that every carbon atom must have FOUR bonds from it

recall the conditions required from your Level 1 Chemistry: high temperature and high pressure with a catalyst

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